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Chapter 12 Stoichiometry 129 SECTION 12.2 CHEMICAL CALCULATIONS (pages 359–366) This section shows you how to construct mole ratios from balanced chemical equations. It then teaches you how to calculate stoichiometric quantities from balanced chemical equations using units of moles, mass, representative particles, and volumes of gases at STP.

### SECTION 12.1 THE ARITHMETIC OF EQUATIONS

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CHEMISTRY READING GUIDE CHAPTER 12 STOICHIOMETRY Name \_\_\_per\_\_ 12.1 What is Stoichiometry? 1. What is meant by the use of the term "quantitative" in the definition of stoichiometry on page 354? 2. How does conservation of mass have to do with stoichiometry? 3.

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Chapter 9: Section 1 - Introduction to Stoichiometry Guided Reading 1. Define: Reaction Stoichiometry - 2. Reaction Stoichiometry is based on \_\_\_ and the Law of \_\_\_\_ \_\_\_\_\_. What do all reaction stoichiometry calculations have to start with? 3.

### Chapter 9: Section 1 Introduction to Stoichiometry Guided ...

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